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Balancing Chemical Equations -Answer Key Balance the equations below: 1)  $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$  2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$  3)  $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$  4)  $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$  5)  $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$  6)  $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$

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Balancing Chemical Equations Worksheet One of the most useful devices for communicating information related to chemical changes is the chemical equation. The equation contains both qualitative and quantitative information related to the nature and quantity of the substances involved in the chemical reaction. It may also include the energy ...

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Balancing Chemical Equations Directions: rewrite and balance the equations below. 1.  $P + O_2 \rightarrow P_2O_5$  \_\_\_\_ 2.  $KClO_3 \rightarrow KCl + O_2$  ... Chemistry IF8766 Page 58 Instructional Fair, Inc. Title: Name: Author: ASD Created Date: 3/7/2011 8:39:23 PM ...

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